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How many atoms  
of hydrogen-1 are  
in a 1.007-g

sample? Recall that  
1.007 amu is the

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mass of one atom  
of hydrogen-1.

Round your answer  
to two significant

digits. 1.007 g 1

H-1 atom  $\frac{1.672$

$10^{-24}$  g a. 6.023

$10^{23}$  H-1 atoms 4.

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contrast

sublimation and

evaporation. In

both processes, the

substances

become a vapor.

During sublimation,

the substance goes

from the solid

phase directly to



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the vapor phase.

During evaporation, particles in a liquid gain enough

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